1. Consider the thermodynamic cycle

$$A \xrightarrow[k_{-1}]{k_{-1}} B$$
$$B \xrightarrow[k_{-2}]{k_{-2}} C$$
$$C \xrightarrow[k_{-3}]{k_{-3}} D$$
$$D \xrightarrow[k_{-4}]{k_{-4}} A$$

,

If this system is able to reach equilibrium, what conditions must the rate constants obey?

2. Show that the sensitivity of the Hill function, defined as

sensitivity 
$$= \frac{d \log y}{d \log x} = \frac{x}{y} \cdot \frac{dy}{dx}$$

is equal to the Hill number n divided by two when x = K, with K being the EC<sub>50</sub>. Thinking about how the sensitivity relates a small change in x,  $\Delta x$ , to a small change in y,  $\Delta y$ ,

$$\frac{\Delta y}{y} = \text{sensitivity} \times \frac{\Delta x}{x}$$

why are Hill functions with a higher n steeper at x = K?