Modelling degradation

Modelling the degradation of a molecule that has a fixed half-life

$$\mathbf{A} \xrightarrow{k} \varnothing$$

The rate equation is

$$\frac{d[A]}{dt} = -k[A]$$

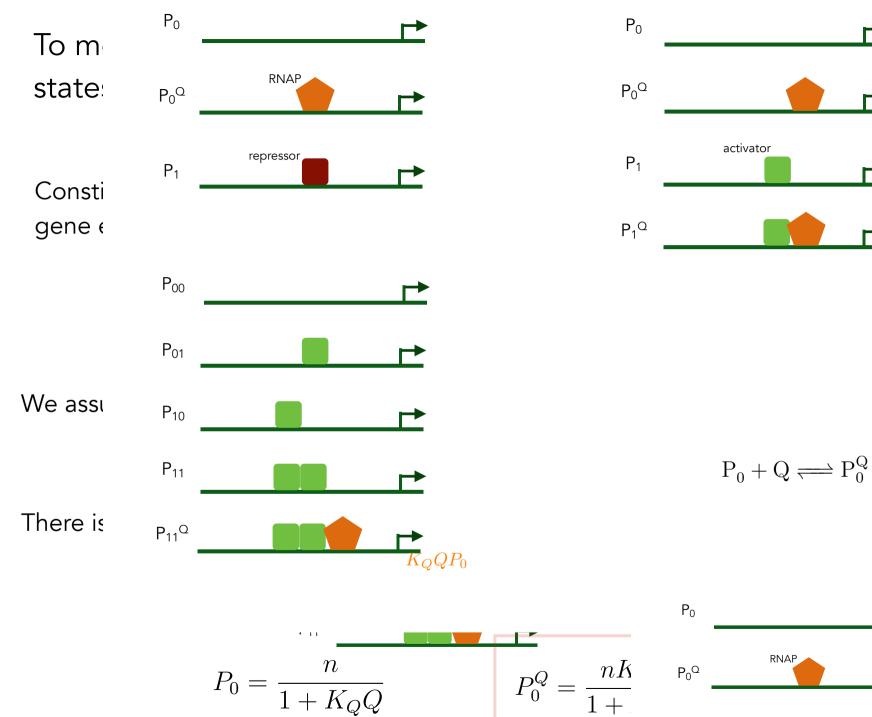
and so

$$[A] = [A]_0 e^{-kt}$$
 or  $[A] = [A]_0 \cdot 2^{\frac{-kt}{\log 2}}$   $using e = 2^{\frac{1}{\log 2}}$ 

The half-life – the time taken for half the molecules to degrade – is inversely proportional to k

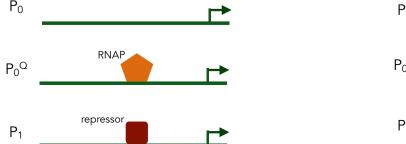
$$t_{\frac{1}{2}} = \frac{\log 2}{k}$$

Modelling gene expression

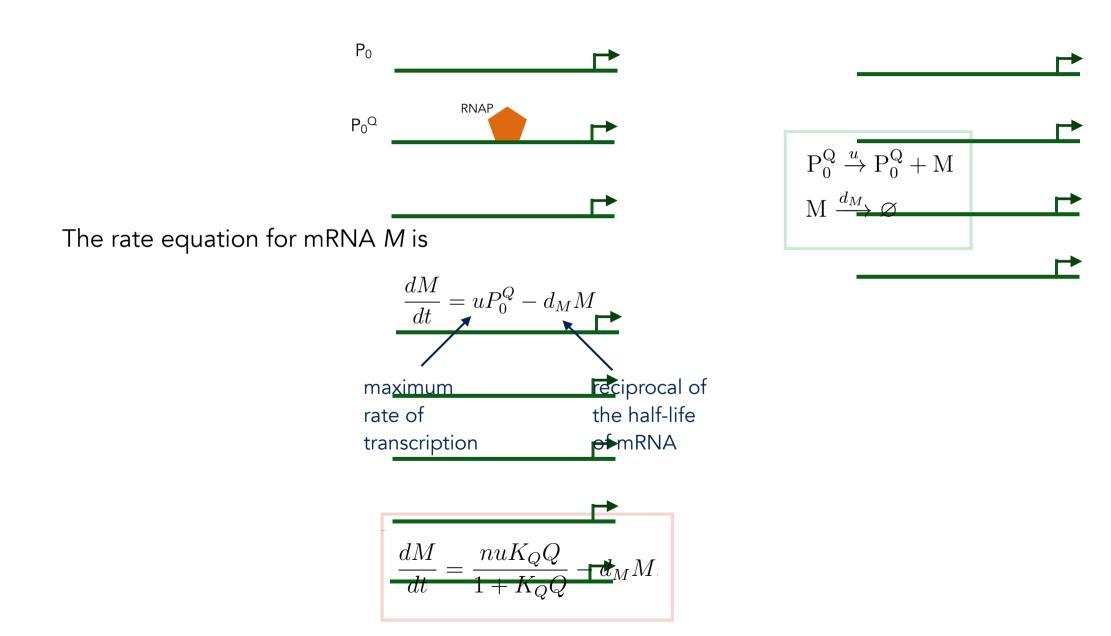


 $P_0^Q =$ 

 $P_0 = \frac{1}{1 + K_Q Q}$ 



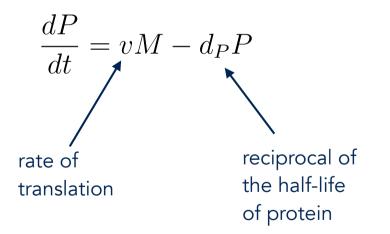
Only the promoter state bound by RNAP initiates transcription



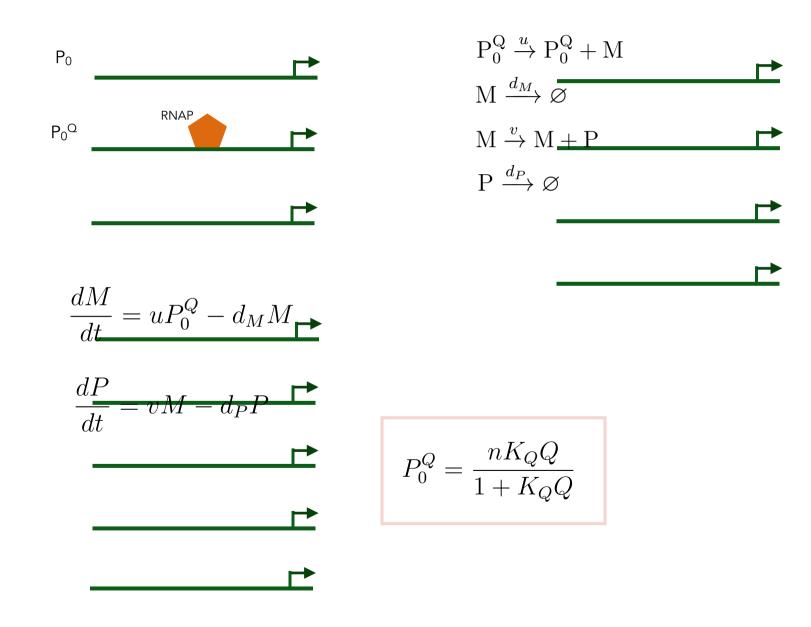
Translation is modelled as a first-order process

$$\begin{array}{l} \mathbf{M} \xrightarrow{v} \mathbf{M} + \mathbf{P} \\ \mathbf{P} \xrightarrow{d_P} \varnothing \end{array}$$

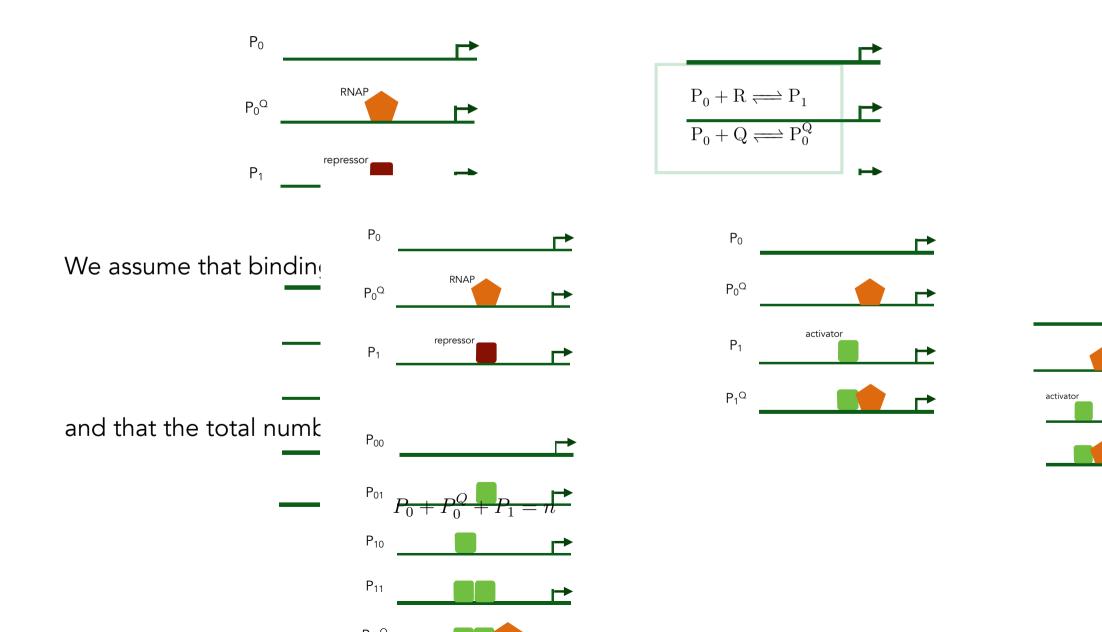
The rate equation for protein *P* is



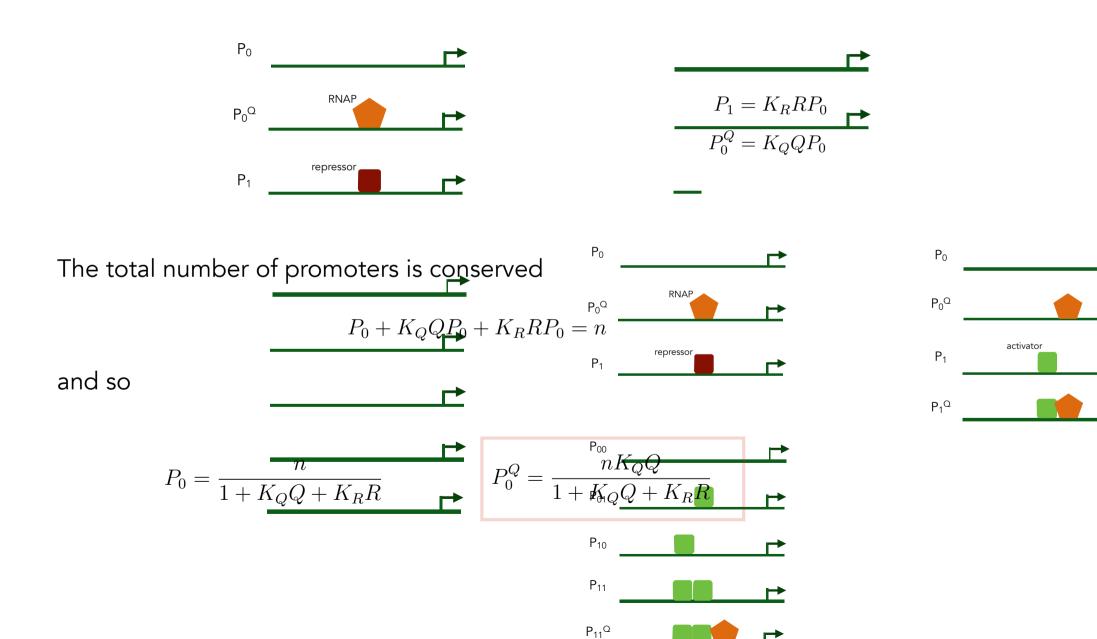
The complete model for a constitutive promoter is then:



Modelling repression by a single repressor competing with RNA polymerase for the promoter



## The higher the number of repressors, the less RNAP binds to the promoter



The model for gene expression from a repressed protein is then

